

LITHIATION OF 6-METHOXY 1,4-BENZODIOXAN:  
FUNCTIONALIZATION AT THE 5-POSITION

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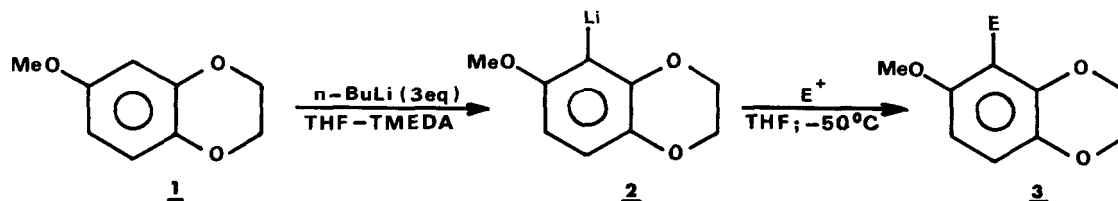
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*Summary:* 6-Methoxy 1,4-benzodioxan 1 undergoes O-lithiation with n-butyllithium to give the intermediate lithio derivative 2. Reaction of 2 with various electrophiles affords 5-substituted 6-methoxy 1,4-benzodioxans 3.

The 1,4-benzodioxan structure is present in many natural products such as neolignans<sup>1</sup> and several synthetic derivatives exhibiting biological activity<sup>2</sup>. However despite extensive work in this area the introduction of substituents on the aromatic nucleus of 1,4-benzodioxans remains somewhat difficult owing to the formation of structural isomers<sup>3</sup>.

In connection with studies directed toward the synthesis of various coumarins, we needed 6-alkoxy and 6-hydroxy 1,4-benzodioxans substituted at the 5 position. We describe here the preparation of such compounds from the 6-methoxy 1,4-benzodioxan 1 readily obtained from the commercially available 6-amino 1,4 benzodioxan<sup>4</sup>.

Deprotonation of alkyl aryl ethers with strong bases is a well established procedure<sup>5</sup>. However the formation of the lithio derivative 2 by treatment of 1 with n-butyllithium is still problematic and requires definite conditions of temperature. Low temperatures (< -50°C) give incomplete deprotonation whereas higher ones (> -30°C) induce a cleavage of the benzodioxan ring<sup>6</sup>. However the treatment of 1 at -50°C with three equivalents of n-butyllithium in tetrahydrofuran in the presence of six equivalents of N,N,N',N'-tetramethylenediamine (TMEDA)<sup>7</sup> readily affords the lithio derivative 2. This carbanion subsequently reacts, at -50°C, with various electrophiles (6 equivalents) to give compounds 3 in fair to excellent yields<sup>8</sup>. The results are summarized in the table.



Replacement of the methoxy group by methoxyethoxymethyl (MEM) or methoxymethyl (MOM) produces an important lowering of the yields.

Table: Preparation of compounds 3

	Electrophile	Yield <sup>(a)</sup>	m.p.	E in product
a	D <sub>2</sub> O	80	oil	D
b	MeI	72	38	Me
c	Me <sub>3</sub> SiCl	92	oil	Me <sub>3</sub> Si
d	Me <sub>2</sub> CO <sup>(b)</sup>	44	oil	Me <sub>2</sub> COH
e	Ph <sub>2</sub> CO	72	143	Ph <sub>2</sub> COH
f	(CH <sub>2</sub> ) <sub>5</sub> CO	82	oil	(CH <sub>2</sub> ) <sub>5</sub> COH
g	Me <sub>2</sub> N-CHO	66 <sup>(c,d)</sup>	124	CHO
h	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>6</sub> CHO	87	oil	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>6</sub> CHOH
i	PhCHO	84	oil	PhCHOH

(a) Yield of pure product isolated by silica gel column chromatography. Not optimized;

(b) 51% of starting material 1 recovered; (c) in those reaction conditions, n-BuLi does not react significantly with the formed aldehyde 3g; (d) N-formyl piperidine as electrophilic reagent gives 60% yield.

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- All compounds 3 gave IR, <sup>1</sup>HNMR and Microanalyses consistent with the assigned structures.

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